Modelling human exposures to plastic additive chemicals and hydrophobic organic contaminants via microplastics

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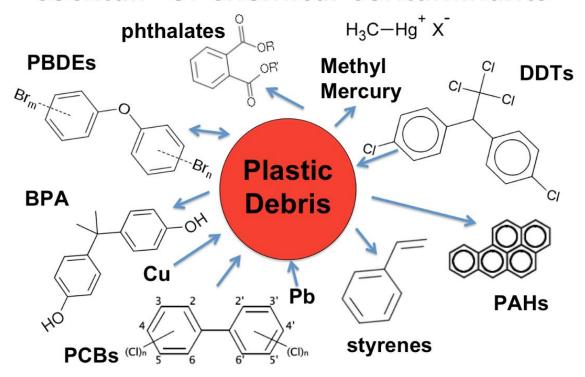






Microplastics as a vector for hydrophobic organic pollutants?

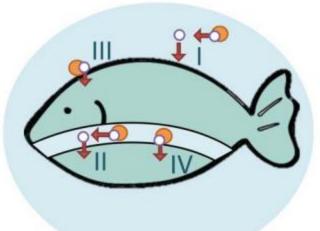
"Cocktail" of Chemical Contaminants





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Plastic Additive Chemicals (PACs)





IPEN's Toxic Plastics Campaign

HOME > PROJECTS

Every stage of the life-cycle of plastic involves toxic chemicals, which threaten human health, the emblodiversity, and the climate.

IPEN's work aims to reveal the toxic threats to health and the environment in each stage of plastics' to:

- · Curb the production of toxic oil, natural gas, and petrochemicals.
- Eliminate and substitute the most toxic chemicals used in the production of plastic
 Strengthen global policies related to plastic waste controls and incineration.
- Promote environmental justice.

TOWARD A PLASTICS TREATY

TOXIC PLASTICS REPORTS

View Media on Toxic Plastics

IPEN is calling for an international plastics treaty that:

- Protects health and the environment
- Ends the production and use of toxic chemicals in plastics
- Removes toxic impacts at all stages of the lifecycle of plastics
- o Bans recycling of plastics containing hazardous chemicals
- Protects the public's right to know about chemicals in plastics and information on plastic production and waste exports
- o Charges plastic producers to finance the treaty
- Promotes safer sustainable materials for a toxics-free circular economy
- o Curbs toxic and climate pollutants

In the meantime, greater transparency on toxic chemical additives used in plastics is needed, along with data on the quantities of plastics made, traded, and disposed of. For more information see our **Plastics Treaty** page.







What is the potential exposure (+ associated health risk) to humans associated with chemical ingestion (PACs & environmental) via MPs?

The "Trojan Horse" Effect

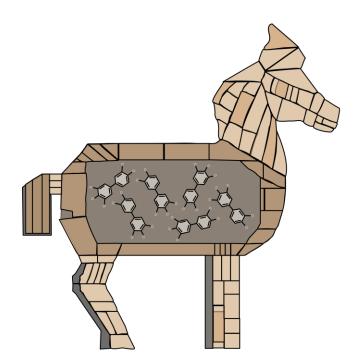


Image: Todd Gouin

MPs unlikely to enhance HOC exposures in marine wildlife BUT what about humans?



ARTICLE

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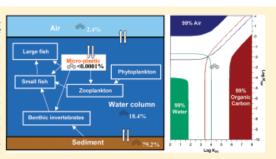
A Thermodynamic Approach for Assessing the Environmental Exposure of Chemicals Absorbed to Microplastic

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Supporting Information

ABSTRACT: The environmental distribution and fate of microplastic in the marine environment represents a potential cause of concern. One aspect is the influence that microplastic may have on enhancing the transport and bioavailability of persistent, bioaccumulative, and toxic substances (PBT). In this study we assess these potential risks using a thermodynamic approach, aiming to prioritize the physicochemical properties of chemicals that are most likely absorbed by microplastic and therefore ingested by biota. Using a multimedia modeling approach, we define a chemical space aimed at improving our understanding of how chemicals partition in the marine environment with varying volume ratios of air/water/organic carbon/polyethylene, where



polyethylene represents a main group of microplastic. Results suggest that chemicals with $\log K_{\rm CW} > 5$ have the potential to partition >1% to polyethylene. Food-web model results suggest that reductions in body burden concentrations for nonpolar organic chemicals are likely to occur for chemicals with $\log K_{\rm CW}$ between 5.5 and 6.5. Thus the relative importance of microplastic as a vector of PBT substances to biological organisms is likely of limited importance, relative to other exposure pathways. Nevertheless, a number of data-gaps are identified, largely associated with improving our understanding of the physical fate of microplastic in the environment.



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Chemical exposures & potential risks from ingesting microplastics in drinking water

Table 3.3 Upper-bound daily intake estimates of chemicals from microplastics, maximum levels of contaminants associated with microplastics, and corresponding MOE

Chemical ^a	Upper bound concentration in microplastic (μg/g)	Maximum daily intake (ng/kg bw/day) ^b	Point of departure (μg/kg bw/day)	
Bisphenol A	0.7297	0.001	609	
Cadmium	3390	5.0	0.8	
Chlordane	0.0144	0.00002	50	Marc
Di(2-ethylhexyl)phthalate	0.0699	0.0001	2500	(MO
Dichlorodiphenyltrichloroethane	7.1	0.0001	1000	5.9
Hexachlorobenzene	0.0587	0.00002	50	1.7
Polyaromatic hydrocarbons	119	0.06	100	2.5
PBDEs	9.9	0.01	100	2.5
PCBs	18.7	0.03	5	1.0 6.0

Margin of exposure (MOE)	Adequacy of MOE	Conclusion	Chemical
5.9 × 10 ⁸	MOE of at least 100	No safety concern	Bisphenol A
1.7×10^{2}	MOE of at least 10 ^c	No safety concern	Cadmium
2.5×10^{9}	MOE of at least 100	No safety concern	Chlordane
2.5×10^{10}	MOE of at least 100	No safety concern	Di(2-ethylhexyl)phthalate
1.0×10^{8}	MOE of at least 100	No safety concern	Dichlorodiphenyltrichloroethane
6.0×10^{8}	MOE of at least 100	No safety concern	Hexachlorobenzene
6.0×10^{5}	MOE of at least 10 000	No safety concern	Polyaromatic hydrocarbons
7.2×10^{6}	MOE of at least 100	No safety concern	PBDEs
1.9×10^5	MOE of at least 1000d	No safety concern	PCBs

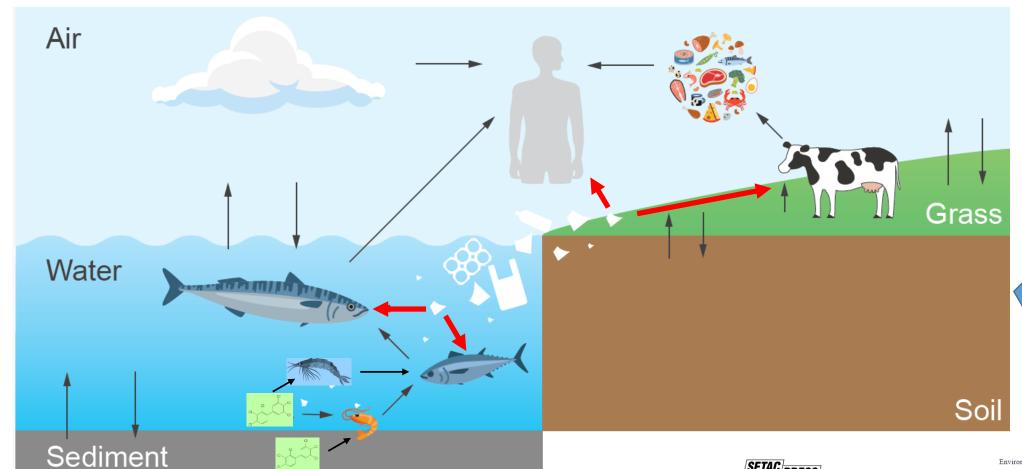




Microplastics in

Modified ACC-Human_{STFADY} Food Web Model





Steady-state, nonequilibrium (Level III) fugacity model for organisms.

Equilibrium assumed for chemicals in environmental media and plastic.



Environmental Toxicology and Chemistry, Vol. 23, No. 10, pp. 2356-2366, 2004

Special Issue Honoring Don Mackay

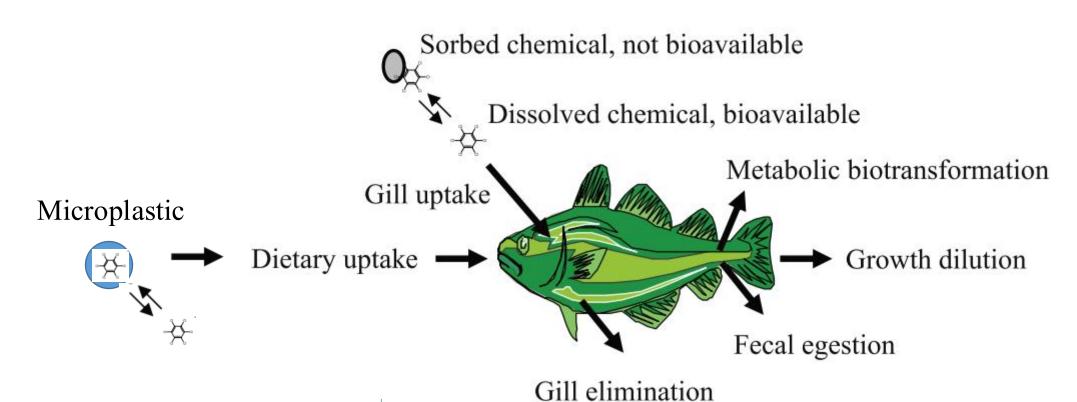
A FOOD CHAIN MODEL TO PREDICT THE LEVELS OF LIPOPHILIC ORGANIC CONTAMINANTS IN HUMANS





Chemical balance in the organism





Integrated Environmental Assessment and Management — Volume 6, Number 2—pp. 210-2 © 2009 SET

Molecular Size Cutoff Criteria for Screening Bioaccumulation Potential: Fact or Fiction?

 $\textit{Jon A Arnot, *} \textit{Michelle I Arnot,$^$} \textit{Don Mackay,$^$} \textit{Yves Couillard,} \textit{§ Drew MacDonald,} \textit{§ Mark Bonnell,} \textit{§ and Pat Doyle} \\ \|$

$$V_i.Z_i \frac{df_i}{dt} = \sum (D_{ji}.f_j) - f_i.\sum D_{ij}$$



Assumptions about HOC behaviour with plastic



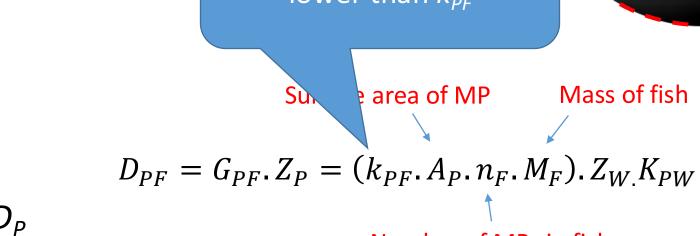
Plastic particles are PE spheres

In equilibrium with sea water ($K_{PF:W}$ from passive sampling)

Two film resistance model for PE-Gut mass transfer

$$\frac{1}{D_{PF}} = \frac{1}{D_P} + \frac{1}{D_F}$$
 k_{FP} assumed to be much

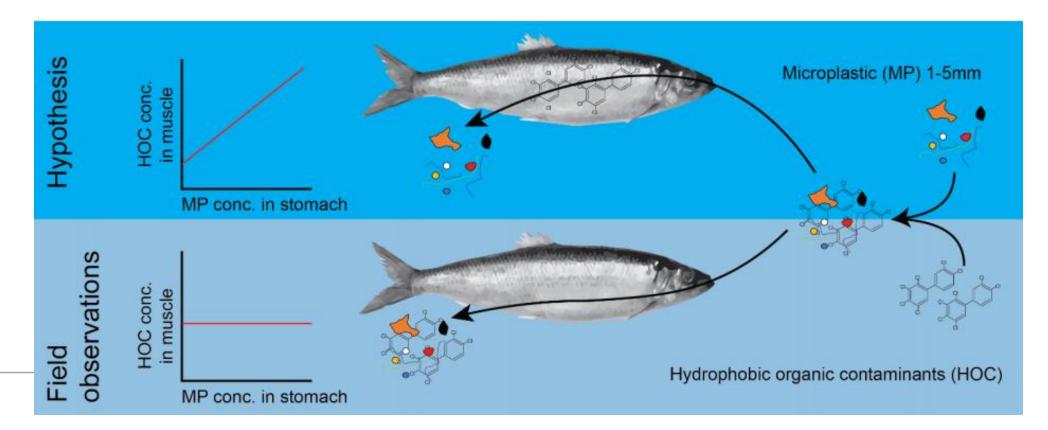
 k_{FP} assumed to be much lower than k_{PF}



Number of MPs in fish

 $Y = 1 \mu \text{m}$ $B_P = 10^{-9} - 10^{-8} \text{ m}^2 \text{ d}^{-1}$ $k_W >>> k_P = 0.1 \text{ m d}^{-1}$ $So \ k_{PF} \approx k_P \ and \ D_{PF} \approx D_P$

Some data suggest HOCs are not desorbed in fish



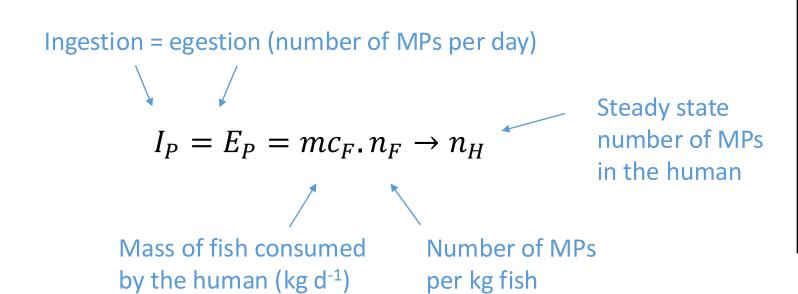


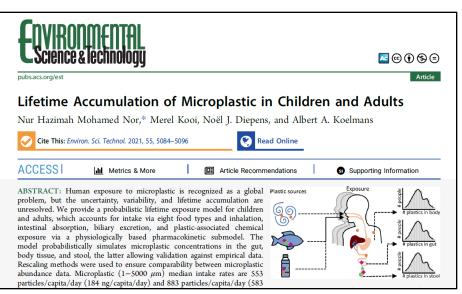
Microplastic Intake, Its Biotic Drivers, and Hydrophobic Organic Contaminant Levels in the Baltic Herring



In Humans







Assume fugacity in plastic (f_P) = fugacity in water (f_W) Steady-state exposure + 1 day residence time in human

Model also enables relative comparison between exposure to hydrophobic organic chemicals via diet and air and what might occur via contaminated MPs



Scenarios



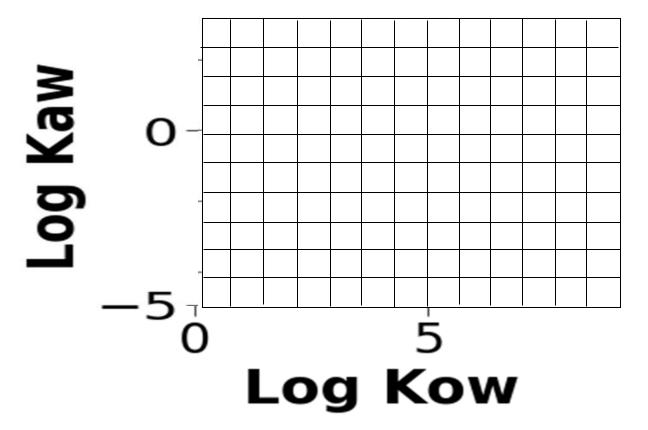
- 1. PAC exposure *only* via ingestion of MPs containing 5% PAC MM⁻¹ by (a) humans; (b) humans & fish; (c) humans, fish & cows
- 2. PAC exposure via the environment* and & MPs, which are ingested by (a) humans; (b) humans & fish; (c) humans, fish & cows
- 3. PAC exposure only via the environment* (no MPs present)



*Environmental exposures (Scenarios 2 & 3) assume all chemicals in all media & in MPs are at equilibrium with sea water (arbitrary conc = 1 ng L^{-1})

Chemical space plots

Hypothetical (persistent) chemicals with varying physicochemical properties



Relative contribution of MP ingestion to total chemical exposure for a given combination of chemical partitioning properties and for given assumptions about MP size and ingestion rate.

100

-80

60

40

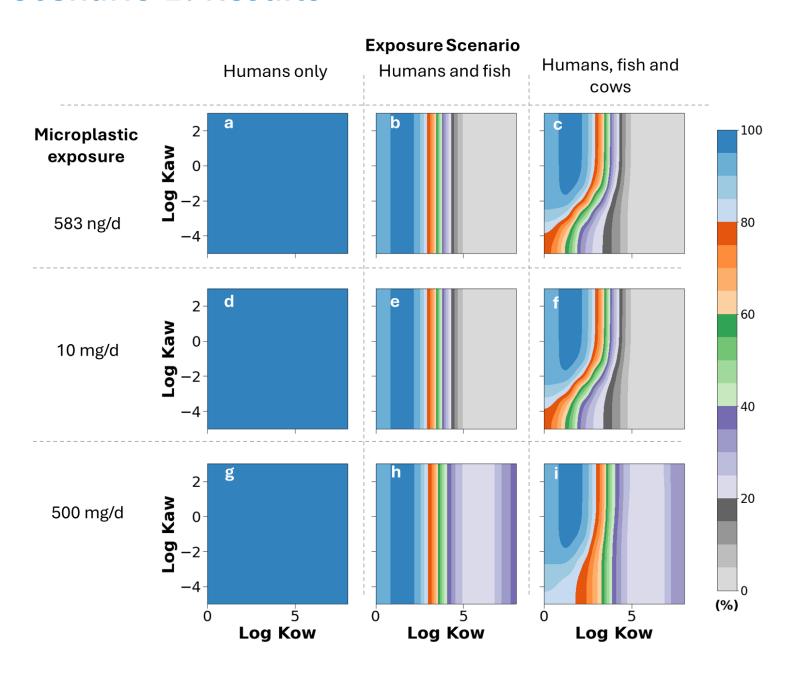
(%)

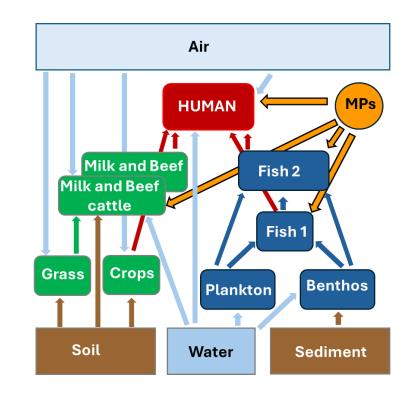
Illustrative examples mainly assume 1 μ m MP size for 3 ingestion rates:

583 ng d⁻¹; 10 mg d⁻¹; 500 mg d⁻¹



Scenario 1: Results





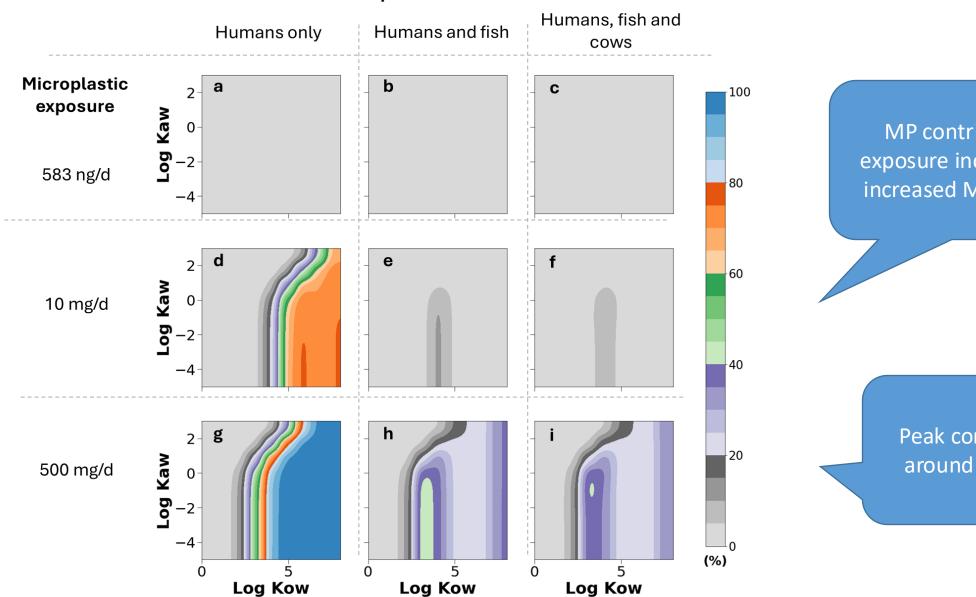
Relative MP contributions to exposure decrease as K_{OW} increases (biomagnification in food webs)



Scenario 2: Results





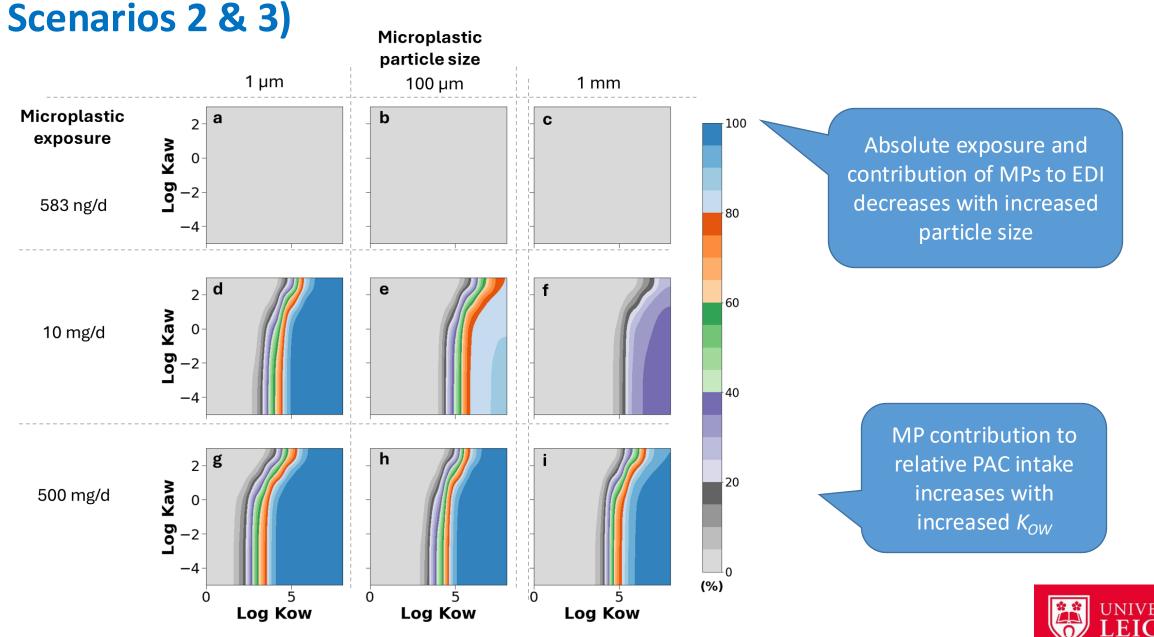


MP contribution to exposure increases with increased MP ingestion

Peak contribution around $log K_{OW} 4$

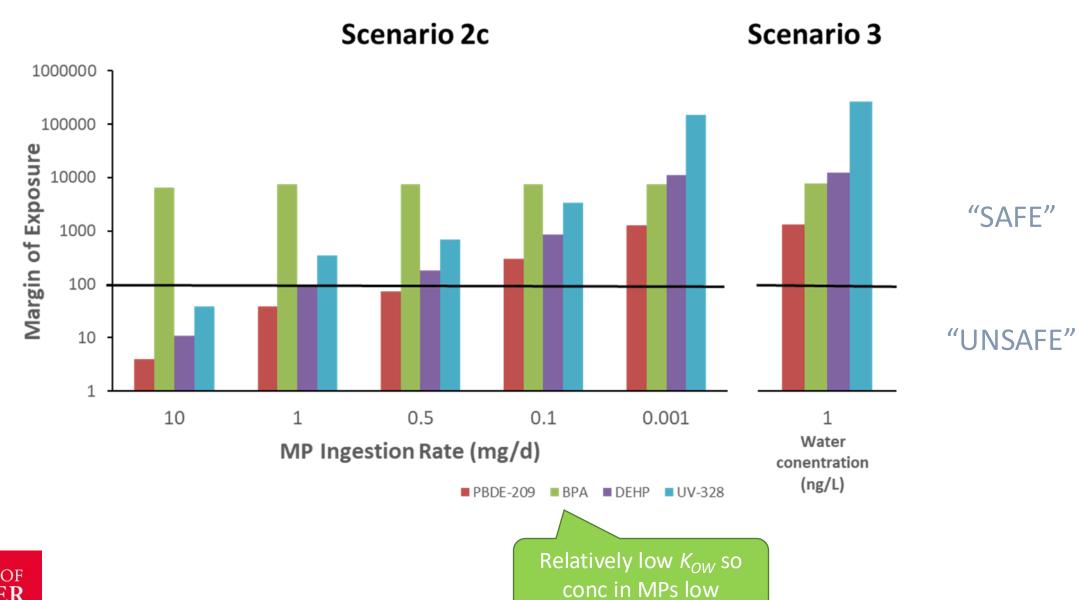


Relative contribution of MPs to PAC EDI (difference between



Application to Risk Assessment for specific PACs







Summary and next steps



- Microplastics could act as vectors for PACs
- Modelling can provide an interpretive framework for
 - Comparing different exposure scenarios
 - Understanding the relative contribution of MPs to total exposure
 - Setting expectations about levels, sensitivity and uncertainty which can support field data collection and interpretation
 - Assessing risks
- Model outputs imply minor contribution of MPs to PAC exposure cf other exposure pathways
- Toxicological thresholds for PACs unlikely exceeded based solely on the ingestion of MP
- UNCERTAINTIES HIGH: Need to represent key inputs as probability distributions in Monte Carlo Simulations

Thanks for Listening!

Acknowledgements

We are grateful to the Centre for Environmental Health and Sustainability at the University of Leicester for pump-priming funding and to the European Union's Horizon 2020 research and innovation programme under Grant agreement No 101003954 (LABPLAS)

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EU Horizon 2020







